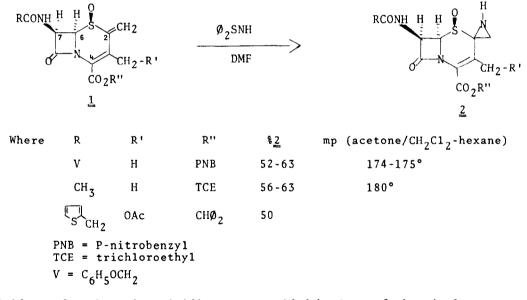
## C<sub>2</sub>-SPIROAZIRIDINO-CEPHALOSPORINS

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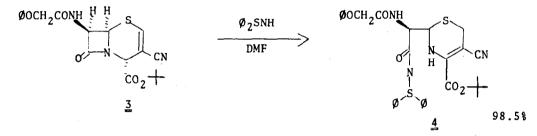
Furukawa and Oae recently reported the Michael type addition of diphenylsulfilimine to  $\alpha$ , $\beta$ -unsaturated ketones and esters to give mixtures of the corresponding aziridine and the enamine<sup>1</sup>.

We now report the use of this reagent with the electrophilic  $C_2$ -exomethylene cephalosporin<sup>2</sup>  $\underline{1}$  to give the  $C_2$ -spiroaziridino-cephalosporin  $\underline{2}$ . Thus a cooled (0°) solution of the diene sulfoxide (1.0 equiv) in DMF was treated with 1.1 equiv of diphenylsulfilimine<sup>3</sup>. After 30 min without cooling the reaction was diluted with ethyl acetate, washed with water, brine, dried, and chromatographed on silica gel to give the corresponding spiroaziridine.

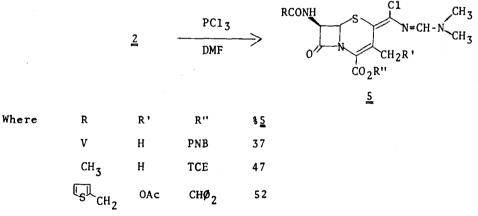


Evidence for the spiroaziridine was provided by loss of the vinyl protons in the NMR (DMSOd<sub>6</sub>) and the presence of a multiplet at 2.6-3.0 $\delta$  for the aziridine methylene and a replaceable proton at 3.4-3.8 $\delta$ . Physical data implies one of two possible geometric isomers for the spiroaziridine. However, the reaction of the N-acetyl diene sulfone  $[R(Me), R'(H), R''(CH_2CCl_3)]$  with diphenylsulfilimine gave a mixture of isomers<sup>4</sup>.

Treatment of other cephem double bonds with  $\emptyset_2$ SNH indicates that under these conditions the  $\Delta^3$ -double bond is inert when the C<sub>3</sub>-substituent is methyl or acetoxymethyl.  $\Delta^2$ -Cephem, with acetoxymethyl at C<sub>3</sub> gave mixtures of  $\Delta^2 - \Delta^3$ , while  $\Delta^2$ -derivatives with electron withdrawing substituents at C<sub>3</sub> resulted in shifting of the double bond and attack of the  $\beta$ -lactam to give the corresponding N-acylsulfilimine<sup>5</sup>.



Attempts to reduce the sulfoxide of  $\frac{2}{2}$  using PC1<sub>3</sub>-DMF<sup>6</sup> gave the chloramidine



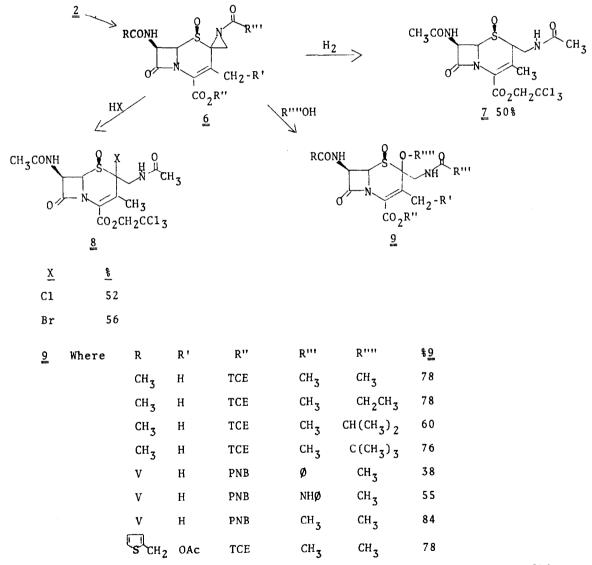
The structure of 5 was evident from the mass spectrum which showed incorporation of chlorine, the IR which showed the  $\beta$ -lactam at 1780 cm<sup>-1</sup> and a strong -N=C- stretch (1630 cm<sup>-1</sup>) and from the NMR (CDCl<sub>3</sub>) which showed a N,N-dimethyl (2.98) and an olefinic proton at 7.48. The conversion of <u>2</u> to <u>5</u> can be rationalized as a nucleophilic attack of the aziridine with dimethyl immonium chloride followed by opening of the aziridine and elimination of HCl. The sulfoxide then undergoes a Pummerer reaction with the Vilsmeier reagent to give the sulfide of the chloramidine<sup>7</sup>. The sulfone aziridine reacts under the same conditions to

<u>5</u>.

No. 41

give the 2'-H(deschloro)amidine.

The spiroaziridine  $\underline{2}$  can be acylated with various acylating agents (CH<sub>3</sub>COC1, ØNCO) to give the spiroacylaziridine <u>6</u> (50-60%, 20%). Thus activated the aziridines undergo reactions with hydrogen to give the amide  $\underline{7}$ , with HX and alcohols to give 2-halo or 2-alkoxy-2-acylaminomethyl derivatives<sup>8</sup> <u>8</u> and <u>9</u>.



Alcoholysis of <u>6</u> results in isomers at  $C_2$  which were separated as the sulfides where R(V), R'(H), R''(PNB),  $R'''(CH_3)$ ,  $R''''(CH_3)$  following  $PBr_3$ -DMF sulfoxide reduction (65%, ratio 4:1).

Both the acid of 5 and the acids of the sulfides of 9 display reduced microbiological activity in comparison to the acid of the corresponding C<sub>2</sub>-methylene derivative.

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